Dissociation Constants for Bases at 25°C

Name	Formula	K_b
Ammonia	NH_3	1.80×10^{-5}
Aniline	**$C_6H_5NH_2$**	**4.30×10^{-10}**
n-Butylamine	$C_4H_9NH_2$	4.0×10^{-4}
***sec*-Butylamine**	**$(CH_3)_2CHCH_2NH_2$**	**3.6×10^{-4}**
tert-Butylamine	$(CH_3)_3CNH_2$	4.8×10^{-4}
Codeine	**$C_{18}H_{21}O_3N$**	**8.91×10^{-7}**
Diethylamine	$(CH_3CH_2)_2NH$	3.09×10^{-5}
Dimethylamine	**$(CH_3)_2NH$**	**5.40×10^{-4}**
Ethylamine	$C_2H_5NH_2$	6.40×10^{-4}
Hydrazine	**H_2NNH_2**	**1.30×10^{-6}**
Hydroxylamine	$HONH_2$	1.10×10^{-8}
Methylamine	**CH_3NH_2**	**4.40×10^{-4}**
Morphine	$C_{17}H_{19}O_3N$	7.41×10^{-7}
Piperidine	**$C_5H_{11}N$**	**1.32×10^{-3}**
Propylamine	$C_3H_7NH_2$	3.5×10^{-4}
Pyridine	**C_5H_5N**	**1.70×10^{-9}**
Quinoline	C_9H_7N	6.31×10^{-10}
Triethanolamine	**$C_6H_{15}O_3N$**	**5.75×10^{-7}**
Triethylamine	$(CH_3CH_2)_3N$	5.25×10^{-4}
Trimethylamine	**$(CH_3)_3N$**	**6.40×10^{-5}**